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© OCR 2015 A Level in Chemistry A General Information . Molar gas volume = 24.0 dm³ mol⁻¹ at room temperature and pressure, RTP Avogadro constant, N_A = 6.02 × 10²³ mol⁻¹ Specific heat capacity of water, c_p = 4.18 J g⁻¹ K⁻¹ Ionic product of water, K_w = 1.00 × 10⁻¹⁴ mol² dm⁻⁶ at 298 K 1 tonne = 10⁶ g Arrhenius equation: $k = Ae^{-E_a/RT}$ or $\ln k = -E_a/RT + \ln A$

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