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Answer: Question 21. In terms of Charles' law explain why -273°C is the lowest possible temperature. Answer: At -273°C , volume of the gas becomes equal to zero, i.e., the gas ceases to exist. Question 22. Critical temperature for CO_2 and CH_4 are 31.1°C and -81.9°C respectively. Which of these has stronger intermolecular forces and why?

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Chapter 3 - Acids, Bases and Salts. Chapter 3 explores the

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properties of acids and bases. The chapter also teaches how different acids and bases react under different circumstances. Additionally, the daily usage of various acids are taught. Chapter 4 - Analytical Chemistry. In Chapter 4, students get acquainted with various salts and solutions.

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Answer Key For Modern Chemistry Study Guide

Chemistry Honors Ch (1-7.1)-Midterm study Guide Choose the best answer from the options that follow each question. ___A 1. A scientist is working in the lab, with compounds containing carbon. She is most likely working in the field of The answer is A because organic chemistry is the study of most carbon-containing compounds. _A___ 2.

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