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Skills: Limiting Reactant
Lab 8 Limiting
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Reactant Stoichiometry -

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2020CHEM 111

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Stoichiometry: Limiting

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Experiment 8 –
Limiting Reactant Pre-
Lab Hints 1a. Refer to
equation 8.1 along with
paragraphs following
equation 8.3. 1b. Refer
to part B of the
procedure. 2. Digesting

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a precipitate by heating
creates larger particles,
which can be more
easily filtered.

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Experiment #8:

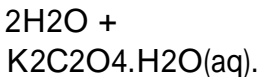
Limiting Reactant

Abstract In chemical
reactions, the

significance of knowing
the limiting reactant is
high. In order to

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increase the percent yield of product, increasing the limiting reactant, possibly, is the most effective. In this experiment we were able to calculate limiting reactants from the reaction of CaCl_2 .



~~Experiment 8 Limiting
Reactant - 1378 Words~~

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— Bartleby

Experiment 8 –

Limiting Reactant Pre-Lab Hints

1a. Refer to equation 8.1 along with paragraphs following equation 8.3.

1b. Refer to part B of the procedure.

2. Digesting a precipitate by heating creates larger particles, which can be more easily filtered. Consider what would happen to

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the smaller particles if
the precipitate was not
digested. 3.

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~~Reactant 10e~~

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Experiment 8 –

Limiting Reactant Pre-

Lab Hints 1a. Refer to

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equation 8.1 along with paragraphs following equation 8.3. 1b. Refer to part B of the procedure. 2. Digesting a precipitate by heating creates larger particles, which can be more easily filtered.

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reagent and then weigh
the show both the

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Explain. The reported mass of the limiting reactant in the original salt mixture will be too high. The mass recorded will be too high if it is not completely dry. This larger mass will result in a larger number when calculating the limiting

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A sample of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O} / \text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$ solid salt mixture is dissolved in ~ 150 mL de-ionized H_2O . The oven dried precipitate has a mass of 0.365 g. The limiting reactant in the salt mixture is

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K₂C₂O₄ H₂O. Prelab

Determine Mass of
K₂C₂O₄ H₂O (aq) in
Salt Mixture in grams.
Do calculation all at
once, round off at the
end.

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The limiting reactant is determined in this experiment.

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Rea ...~~

Lab 8 Limiting
Reactant Name: Tashfia
Hasan Lab Partner:

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Hayley Karr, Natalie

Sondhi, Tina Takla,

Becca, Som Keshav

Course: E01, Tuesdays

6 PM - 8:45 PM

Instructor Name:

Obiajulu Nwanze

Laboratory Assistant

Name: Camila

Gonzalez Date

Experiment was

Performed: September

22, 2020. Prelab

Questions:

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~~Lab 8 Limiting~~

~~Reactant - CHEM 1300~~

~~General Chemistry I ...~~

Question: Exp 8:

Synthesis Of Alum

Drawer #: Name:

Prelab Questions: 2 Al

$(s) + 2 \text{KOH}(aq) + 4 \text{H}_2\text{SO}_4$

$(aq) + 22 \text{H}_2\text{O}$

$2\text{KAl}(\text{SO}_4)_3 \cdot 12\text{H}_2\text{O}(s) + 3$

$\text{H}_2(g)$ 1) What Is The

Molar Mass Of

$\text{KAl}(\text{SO}_4)_3 \cdot 12\text{H}_2\text{O}$?

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Exp 8 Limiting

(Remember To Add 12
H₂O Molecules I.e. Add
24 Extra Hydrogens

And 12 Extra Oxygens)

2) Calculate The Mass
Of Alum That Can Be
Produced From 14.471g
Of Al Grams Almoles Al
Moles "alum" ...

~~Solved: Exp 8: Synthesis~~

~~Of Alum Drawer #:~~

~~Name: Prelab Qu ...~~

The percent

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Exp 8 Limiting

Composition of the salt is

determined by first

testing for the limiting

reactant. The limiting

reactant for the

formation of solid

calcium monohydrate is

determined from two

precipitation tests of the

final mixture from Part

A. 1. Mixture is tested

for an excess of oxalate

ion with limited calcium

ion 2.

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CHEM 1510L

Experiment 004

Limiting Reagent and

Percent Yield ... Pre-lab

write-up instructions ...

Introduction to Limiting

Reactant and Excess

Reactant - Duration:

16:58. Tyler DeWitt ...

~~CHEM 111 limiting~~

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The limiting reactant is related to the product using the stoichiometry of the balanced equation. In the example above, since Cl_2 is the limiting reactant and it could form 188.1 g of AlCl_3 product, that will be the theoretical yield for the reaction. Of course, in practice this

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~~Experiment 3 Limiting Reactants~~

dessaniel jaquez

10/18/17 experiment

limiting reactant co-

workers: zach, sophie

objectives: to determine

the limiting reactant in

mixture of two soluble

salts. Sign in Register;

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