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Relationships  
Stoichiometry  
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*Mole Lab*

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The Mole **Lab Experiment #7:**

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**The Stoichiometry of a  
Chemical Reaction. Target  
Stoichiometry Lab** Target

Mole Lab ~~Avogadro's Number,  
the Mole and How to Use the  
Mole~~

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Flinn ChemTopic Labs ~~The Mole  
Concept~~ Introduction to

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~~Moles Molar Volume of a Gas  
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Target Gas Law Lab Volume of  
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(Not the animal, the other  
one.) - Daniel Dulek How to  
Collect and Measure Gases  
Stoichiometry lab Na<sub>2</sub>CO<sub>3</sub> to~~

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NaCl AChem - Lab -

Stoichiometric Relationships  
in a Chemical Reaction

Hydrogen Peroxide 30%

~~Stoichiometry Tutorial: Step  
by Step Video + review~~

~~problems explained | Crash~~

~~Chemistry Academy~~ **Molar**



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**Volume of Hydrogen Gas - Mg**

**+ HCl** ~~What is a mole~~

Chemistry: What is the Mole

(Avogadro's Number)? 2

practice problems | Homework

Tutor Very Common Mole

Questions How to Calculate

Molar Heat of Solution -

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Sample Problem *Determination  
of the molar Volume of a Gas  
at STP Converting Between  
Moles and Liters of a Gas at  
STP* ~~Mole Concept~~

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Chemistry Lesson: Mole  
Calculations ICHm0085 molar  
volume of hydrogen gas lab

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~~Course Chemistry #18 Models~~

~~to Illustrate Ionic and~~

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**Relationships**

Fundamental in chemistry,

the mole concept allows

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access the submicroscopic world of atoms and molecules. Understanding and applying the mole concept makes possible the analysis of compounds and their chemical formulas while studying the properties of

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With the Mole  
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Activity, engage in a  
collection of activities  
that help students visualize  
the size of a mole,  
understand the meaning of



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relationships behind the calculations, dispel molar misconceptions, effectively calculate molar mass and determine the identity of a mystery compound based on molar mass calculations. Available as part of the

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Stoichiometry Introduction

Stoichiometry Balloon Races

Limiting and Excess

Reactants Most stoichiometry calculations in the classroom are performed using exact (stoichiometric) mole ratios of reactants and

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products.

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Materials for Molar Samples  
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Assessment are available

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Calcium, Turnings C0203

Charcoal, Block C0146

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Copper, Wire I0011 Iron,  
Filings L0090 Lead, Shot  
M0001 Magnesium, Ribbon

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molecules. Understanding and  
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My "go to" procedure for this lab is found in ChemTopic Labs Volume 7: Molar Relationships & Stoichiometry by Flinn. In Flinn's version of the lab, there are detailed instructions to heat the

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magnesium with the lid on,  
then open the lid every  
three minutes, and then heat  
it some more with the lid at  
a tilt. All of this  
manipulation of the crucible  
lid invariably results in at  
least one dropped and broken

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Relationships  
lid each lab period.

Stoichiometry Answers

**Blow the Lid Off of  
Magnesium Oxide Lab - The  
Art of ...**

To find the volume of gas  
that will be used to  
calculate the molar volume,

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the process of water displacement can be used.  
Reference Citation. Cesa, J. (2002). ChemTopic labs: Experiments and demonstrations in chemistry (vol. 9). Batavia, Il: Flinn Scientific. Calculations

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**Molar Volume of Hydrogen Lab  
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know where i can find the  
answers to the pre-lab

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ChemTopic Labs on magnesium  
oxide? A piece of iron

weighing 85.65g was burned  
in air. The mass of the iron  
oxide produced was 118.37g

!. Use the law of  
conservation of mass to



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calculate the mass of oxygen  
that reacted with iron.

**lab!: Magnesium oxide I  
REALLY NEED ANSWERSSSSS ...**

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## **Flinn Chemtopic Labs Organic Compounds Answers**

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guaranteed to help you teach chemistry. Made with the guidance of the best chemistry teachers in America.

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The reaction of sodium bicarbonate with acetic acid generates carbon dioxide gas. The reactants combine in a 1:1 mole ratio. The theoretical number of moles of gas for each reaction and the limiting and excess

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relationships in each case are summarized in the Results Table.  $\text{NaHCO}_3 (\text{s}) + \text{CH}_3\text{CO}_2\text{H} (\text{aq}) \rightarrow \text{NaCH}_3\text{CO}_2 (\text{aq}) + \text{CO}_2 (\text{g}) + \text{H}_2\text{O} (\text{l})$  Results Table Flask

## **Stoichiometry Balloon Races**

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Stoichiometry . Page 3

-Micro Mole Rockets 4.

Prepare two hydrogen gas  
generators by placing about  
four pieces of mossy zinc

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into the bottom of two small test tubes. 5. Prepare two oxygen gas generators by placing about 2 mL of yeast suspension into the

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and Oxygen Mole Ratio As

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Relationships &

Stoichiometry Introduction

The combustion reaction of hydrogen and oxygen is used to produce the explosive

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energy needed to power the  
space shuttle. The reaction  
is also being engineered to  
serve as a source of  
continuous energy

**Page I Micro Mole Rockets -  
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The combustion reaction of hydrogen and oxygen is used to produce the explosive

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energy needed to power the space shuttle. The reaction is also being engineered to serve as a source of continuous energy for fuel cells in electric vehicles.

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Analyze the formulas of compounds and determine molar relationships among

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relationships create  
nanostructures, in addition  
to the experimental  
techniques used to  
characterize them, as well  
as some of the surprising  
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chapters cover every

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minimize exposures.

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