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INTRODUCTION TO MASS SPECTROMETRY Instrumentation, Applications and Strategies for Data Interpretation FOURTH EDITION J. THROCK WATSON Professor of Biochemistry and of Chemistry Michigan State University East Lansing, Michigan O. DAVID SPARKMAN Adjunct Professor of Chemistry College of the Pacific University of the Pacific Stockton, California

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Mass spectrometry (MS) is a powerful analytical technique widely used by chemists, biologists, medical researchers, and environmental and forensic scientists, among others. With MS, we are looking at the mass of a molecule, or of different fragments of that molecule. The basics of a mass spectrometry experiment

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Instead, mass spectrometry is based upon the motion of a charged particle, called an ion, in an electric or magnetic field. The mass to charge ratio (m/z) of the ion effects this motion. Since the charge of an ** electron is known, the mass to charge ratio a measurement of an ion's mass. Typical mass spectrometry research focuses on the

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INSTRUMENTATION STATUS: all instrumentation are in the operation Mass spectrometry laboratory comprise of the state of art industry-leading instrumentations, available to use for staff and students. The laboratory is housed in LG11 in UCL Chemistry Building. The laboratory deals with more than 30,000 different types of samples per year.

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Mass spectrometry A mass spectrometer is simply a device designed to determine the mass of individual atoms or molecules. Atoms of different elements have different masses and thus a knowledge of the molecular mass can very often be translated into a knowledge of the chemical species involved.

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