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_____ 1) ammonium sulfite % N 24.1 % H 6.9 % S 27.6 % O 41.3

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_____ 2) aluminum acetate % Al 13.2 % C 35.3 Percent
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moles = mass (in grams) divided by molecular weight. or: moles = g / MW therefore, by using algebra: $g = \text{moles} \times \text{MW}$ or $\text{MW} = g /$

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Answered: Describe the difference between Gram(-)... | bartleby

Answers . 1. 1.587 grams per milliliter 2. 407.5 grams 3. 559
milliliter 4. 1051.2 grams 5. 26561 grams or 26.56 kilograms 6. a.

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7.8 grams per milliliter b. 3×10^{-2} kilograms/centimeter 3 7. Liquid
A. (0.75 grams per milliliter) 8. 68 kilograms 9. 8.33 pounds (2.2
kilograms = 1 pound, 1 liter = 0.264 gallons) 10. 483.6 centimeters³

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