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solid into gas without changing into liquid is known as _____. A) Heating done clear. B) Evaporation done clear. C) Melting ...

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(i) It is an inflammable gas (ii) It is the lightest gas. (iii) It is not easily

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liquefied. (iv) It is a strong oxidizing agent. (e) For the reaction $\text{PbO} + \text{H}_2 \rightarrow \text{Pb} + \text{H}_2\text{O}$, which of the following statements is wrong? i) H_2 is the reducing agent. (ii) PbO is the oxidizing agent. (iii) PbO is oxidized to Pb (iv) H_2 is oxidized to H_2O .

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Study the following figure and answer questions. a) After heating Calcium carbonate, which gas is formed in a test tube? b) When we pass this gas through limewater what change, did you observe? c)

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Write down the chemical reaction showing the product formation after heating the Calcium carbonate.

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