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Harold Davis [NEW]

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Bonds Use with Chapter

7, Section 7.2 1. How

many valence electrons

does a neutral

magnesium (Mg) atom

have? 2. What is the

charge on a magnesium

ion? What does

magnesium have to do

to form such an ion, and

why does it tend to do

so? 3. How many

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valence electrons does a
single neutral chlorine
(Cl atom have? 4.

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For each of the following pairs of elements, underline the one that would replace the other element in a compound.

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Considering your
answer to question 5,
how many electrons can
fit in the entire 5t energy

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level? can (A Vel is a ha
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notation for an electron
spinning clockwise in a
p sublevel in the 4th ...

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WORKSHEET Atomic
Orbitals 1. What is the
shape of an s orbital?

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Chapter 5, Section 5.2 2.

What is the relationship
between the size of an s

orbital and the principal
energy level in which it

is found? 3. What is the
shape of a p orbital?

How many p orbitals are
there in a ublevel? 4.

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road, and also

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22 – Formation of Ions
1. calcium and oxygen
2. Yes; each contains
equal numbers of
protons and electrons. 3.
ionization energy 4.

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positive; cation 5.
negative; anion 6. No; it
is not a stable octet of
electrons. 7.

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